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## Key indicators

Single-crystal X-ray study
$T=295 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.004 \AA$
$R$ factor $=0.046$
$w R$ factor $=0.143$
Data-to-parameter ratio $=16.4$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## ( $N$-Hydroxy- $N$-phenylbenzamidato- $\kappa^{2} O, O^{\prime}$ )oxo[salicylaldehyde (4-nitrobenzoyl)hydrazon-ato- $\left.\kappa^{3} O, N, O^{\prime}\right]$ vanadium $(\mathrm{V})$

The vanadium $(\mathrm{V})$ atom in $\left[\mathrm{V}\left(\mathrm{C}_{14} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}_{4}\right)\left(\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{NO}_{2}\right) \mathrm{O}\right]$ is coordinated by an oxo O atom, the N and two O atoms of the tridentate hydrazonate, and the two O atoms of the bidentate hydroxamate in a distorted octahedral geometry.

## Comment

Bis(acetylacetonato)vanadium(IV) reacts with $N$-benzoyl- $N$ phenylhydroxylamine and benzoylacetone 4-methoxybenzoylhydrazone to form (benzoylacetone 4-methoxybenzoylhydrazonato)( $N$-benzoylhydroxylaminato)oxovanadium(IV) (Gao et al., 2005). A similar reaction with salicylaldehyde(4-nitrobenzoyl)hydrazone, which has an acidic H atom in the form of a hydroxy group, furnished the corresponding compound, but the metal atom has been oxidized to the $5+$ oxidation state, (I). The $\mathrm{V}^{\mathrm{V}}$ atom is coordinated by an oxo O atom, one N and two O atoms of the tridentate hydrazonate, and the two O atoms of the bidentate hydroxamate in a distorted octahedral geometry.

(I)

## Experimental

Salicylaldehyde(4-nitrophenoxyacetyl)hydrazone was synthesized by condensing salicylaldehyde with an equimolar quantity of (4-nitrobenzoyl)hydrazine in ethanol. An ethanol solution ( 15 ml ) of $\mathrm{VO}(\mathrm{acac})_{2}(0.26 \mathrm{~g}, 2 \mathrm{mmol})$ was added to an ethanol solution ( 15 ml ) of salicylaldehyde(4-nitrobenzoyl)hydrazone ( $0.56 \mathrm{~g}, 2 \mathrm{mmol}$ ). The mixture was heated for 1 h . The addition of an ethanol solution ( 15 ml ) of $N$-benzoyl- $N$-phenylhydroxylamine ( $0.41 \mathrm{~g}, 2 \mathrm{mmol}$ ) gave a red solution. The mixture was filtered, and the dark red compound was collected and then dissolved in dichloromethane. Crystals were obtained by the diffusion of $n$-hexane into the filtrate over a period of several days. Analysis found: C 57.62, H 3.47, N $9.99 \%$; calculated for $\mathrm{C}_{27} \mathrm{H}_{19} \mathrm{~N}_{4} \mathrm{O}_{7}$ V: C 57.66, H 3.41, N $9.96 \%$.

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Figure 1
ORTEPIII (Burnett \& Johnson, 1996) molecular view of (I), with the atom-labelling scheme. Displacement ellipsoids are drawn at the $30 \%$ probability level. H atoms have been omitted for clarity.

## Crystal data

| [ $\mathrm{V}\left(\mathrm{C}_{14} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}_{4}\right)\left(\mathrm{C}_{13} \mathrm{HNO}_{3}\right) \mathrm{O}$ ] | $Z=2$ |
| :---: | :---: |
| $M_{r}=562.40$ | $D_{x}=1.465 \mathrm{Mg} \mathrm{m}^{-3}$ |
| Triclinic, $P \overline{1}$ | Mo $K \alpha$ radiation |
| $a=10.263$ (2) $\AA$ | Cell parameters from 11507 |
| $b=10.737$ (2) $\AA$ | reflections |
| $c=12.623$ (3) $\AA$ | $\theta=3.2-27.5^{\circ}$ |
| $\alpha=86.48$ (3) ${ }^{\circ}$ | $\mu=0.44 \mathrm{~mm}^{-1}$ |
| $\beta=86.20$ (3) ${ }^{\circ}$ | $T=295$ (2) K |
| $\gamma=66.81(3)^{\circ}$ | Prism, red |
| $V=1274.8$ (5) $\AA^{3}$ | $0.38 \times 0.26 \times 0.18 \mathrm{~mm}$ |
| Data collection |  |
| Rigaku R-AXIS RAPID IP diffractometer | 5774 independent reflections 4630 reflections with $I>2 \sigma(I)$ |
| $\omega$ scans | $R_{\text {int }}=0.013$ |
| Absorption correction: multi-scan | $\theta_{\text {max }}=27.5^{\circ}$ |
| (ABSCOR; Higashi, 1995) | $h=-13 \rightarrow 12$ |
| $T_{\text {min }}=0.842, T_{\text {max }}=0.923$ | $k=-13 \rightarrow 13$ |
| 12599 measured reflections | $l=-16 \rightarrow 16$ |

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0859 P)^{2}\right. \\
& \quad+0.1841 P] \\
& \quad \text { where } P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.65 \text { e } \AA^{-3} \\
& \Delta \rho_{\min }=-0.24 \mathrm{e}^{-3} \\
& \text { Extinction correction: none }
\end{aligned}
$$

$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.046$
$w R\left(F^{2}\right)=0.143$
$S=1.09$
5774 reflections
352 parameters
H -atom parameters constrained

The $\mathrm{C}-\mathrm{C}$ distances in aromatic rings were restrained to be equal with an effective standard deviation of 0.01 , as the free refinement led to a wide spread of these distances. Some of the anisotropic displacement parameters have large components, suggesting the occurrence of disorder; however, any attempts to construct chemically reasonable models failed. H atoms were placed in calculated positions $(\mathrm{C}-\mathrm{H}=0.93 \AA)$ and were allowed to ride on their parent C atoms, with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\mathrm{eq}}(\mathrm{C})$.

Data collection: RAPID-AUTO (Rigaku Corporation, 1998); cell refinement: RAPID-AUTO; data reduction: CrystalStructure (Rigaku/MSC, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPIII (Burnett \& Johnson, 1996) and ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: SHELXL97.

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