metal-organic papers

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Key indicators

Single-crystal X-ray study T = 295 KMean σ (C–C) = 0.004 Å R factor = 0.046 wR factor = 0.143 Data-to-parameter ratio = 16.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

(*N*-Hydroxy-*N*-phenylbenzamidato- $\kappa^2 O, O'$)oxo[salicylaldehyde (4-nitrobenzoyl)hydrazonato- $\kappa^3 O, N, O'$]vanadium(V)

The vanadium(V) atom in $[V(C_{14}H_9N_3O_4)(C_{13}H_{10}NO_2)O]$ is coordinated by an oxo O atom, the N and two O atoms of the tridentate hydrazonate, and the two O atoms of the bidentate hydroxamate in a distorted octahedral geometry. Received 3 March 2005 Accepted 18 March 2005 Online 25 March 2005

Comment

Bis(acetylacetonato)vanadium(IV) reacts with *N*-benzoyl-*N*-phenylhydroxylamine and benzoylacetone 4-methoxybenzoylhydrazone to form (benzoylacetone 4-methoxybenzoylhydrazonato)(*N*-benzoylhydroxylaminato)oxovanadium(IV) (Gao *et al.*, 2005). A similar reaction with salicylaldehyde(4-nitrobenzoyl)hydrazone, which has an acidic H atom in the form of a hydroxy group, furnished the corresponding compound, but the metal atom has been oxidized to the 5+ oxidation state, (I). The V^V atom is coordinated by an oxo O atom, one N and two O atoms of the tridentate hydrazonate, and the two O atoms of the bidentate hydroxamate in a distorted octahedral geometry.



Experimental

Salicylaldehyde(4-nitrophenoxyacetyl)hydrazone was synthesized by condensing salicylaldehyde with an equimolar quantity of (4-nitrobenzoyl)hydrazine in ethanol. An ethanol solution (15 ml) of VO(acac)₂ (0.26 g, 2 mmol) was added to an ethanol solution (15 ml) of salicylaldehyde(4-nitrobenzoyl)hydrazone (0.56 g, 2 mmol). The mixture was heated for 1 h. The addition of an ethanol solution (15 ml) of N-benzoyl-N-phenylhydroxylamine (0.41 g, 2 mmol) gave a red solution. The mixture was filtered, and the dark red compound was collected and then dissolved in dichloromethane. Crystals were obtained by the diffusion of *n*-hexane into the filtrate over a period of several days. Analysis found: C 57.62, H 3.47, N 9.99%; calculated for $C_{27}H_{19}N_4O_7V$: C 57.66, H 3.41, N 9.96%.

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Figure 1

ORTEPIII (Burnett & Johnson, 1996) molecular view of (I), with the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. H atoms have been omitted for clarity.

Crystal data

[V(C ₁₄ H ₉ N ₃ O ₄)(C ₁₃ HNO ₃)O]	Z = 2
$M_r = 562.40$	$D_x = 1.465 \text{ Mg m}^{-3}$
Triclinic, $P\overline{1}$	Mo $K\alpha$ radiation
a = 10.263 (2) Å	Cell parameters from 11 507
b = 10.737 (2) Å	reflections
c = 12.623 (3) Å	$\theta = 3.2-27.5^{\circ}$
$\alpha = 86.48 \ (3)^{\circ}$	$\mu = 0.44 \text{ mm}^{-1}$
$\beta = 86.20 \ (3)^{\circ}$	T = 295 (2) K
$\gamma = 66.81 \ (3)^{\circ}$	Prism, red
V = 1274.8 (5) Å ³	$0.38 \times 0.26 \times 0.18 \text{ mm}$
Data collection	
Rigaku R-AXIS RAPID IP	5774 independent reflections

Rigaku R-AXIS RAPID IP diffractometer ω scans Absorption correction: multi-scan (*ABSCOR*; Higashi, 1995) $T_{min} = 0.842, T_{max} = 0.923$ 12 599 measured reflections 5774 independent reflections 4630 reflections with $I > 2\sigma(I)$ $R_{int} = 0.013$ $\theta_{max} = 27.5^{\circ}$ $h = -13 \rightarrow 12$ $k = -13 \rightarrow 13$ $l = -16 \rightarrow 16$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0859P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.046$	+ 0.1841P]
$vR(F^2) = 0.143$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.09	$(\Delta/\sigma)_{\rm max} = 0.001$
5774 reflections	$\Delta \rho_{\rm max} = 0.65 \ {\rm e} \ {\rm \AA}^{-3}$
352 parameters	$\Delta \rho_{\rm min} = -0.24 \text{ e} \text{ \AA}^{-3}$
H-atom parameters constrained	Extinction correction: none

The C–C distances in aromatic rings were restrained to be equal with an effective standard deviation of 0.01, as the free refinement led to a wide spread of these distances. Some of the anisotropic displacement parameters have large components, suggesting the occurrence of disorder; however, any attempts to construct chemically reasonable models failed. H atoms were placed in calculated positions (C–H = 0.93 Å) and were allowed to ride on their parent C atoms, with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$.

Data collection: *RAPID-AUTO* (Rigaku Corporation, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSC, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP1II* (Burnett & Johnson, 1996) and *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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